

Process Development and Scale up of Advanced Active Battery Materials – Gradient Cathode Materials

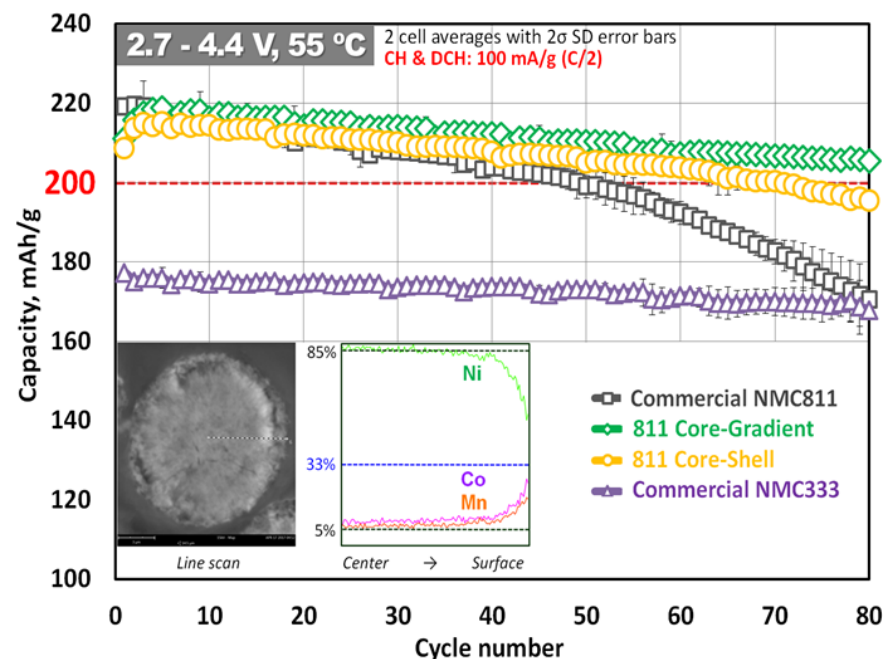
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Project ID: ES167

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Overview

Timeline

- Project start date: Oct. 2010
- Project end date: Sept. 2017
- Percent complete: on going

Budget

- Total project funding:
 - \$1.2M in FY16
 - \$1.1M in FY17

Barriers

- Cost: Reduce manufacturing costs with advanced, continuous processing methods
- Performance: Optimization of particle structure and composition combination for maximum gradient material performance

Partners

- Active material process R&D:
 - Argonne's Applied R&D Group
 - Material synthesis and scale-up
 - University of Illinois at Chicago
 - 3D elemental mapping
 - Technische Universität Braunschweig
 - Particle stress study
 - Brookhaven National Laboratory, EES, Dr. Xiao-Qing Yang's group
 - Thermal stability
 - Laminar
 - Taylor Vortex Reactor process scale-up

Objectives - Relevance

- The objective of this program is to carry out a systematic research to:
 - Synthesize and evaluate various types of concentrated gradient cathode materials targeting **> 220 mAh/g with > 95% capacity retention** after 100 cycles.
 - Develop cost-effective batch and **continuous processes** for the scale-up of concentrated gradient cathode materials.
 - Provide **sufficient quantities** of these materials produced under rigorous quality control specifications for industrial evaluation of further research.

- The relevance of this program to the DOE Vehicle Technologies Program is:
 - The program is a key missing link between discovery of advanced battery materials, market evaluation of these materials and high-volume manufacturing.
 - Reducing the risk associated with the commercialization of new battery materials.
 - This program provides large quantities of materials with consistent quality.
 - For industrial validation in large format prototype cells.
 - For further research on concentrated gradient cathode materials.

Milestones

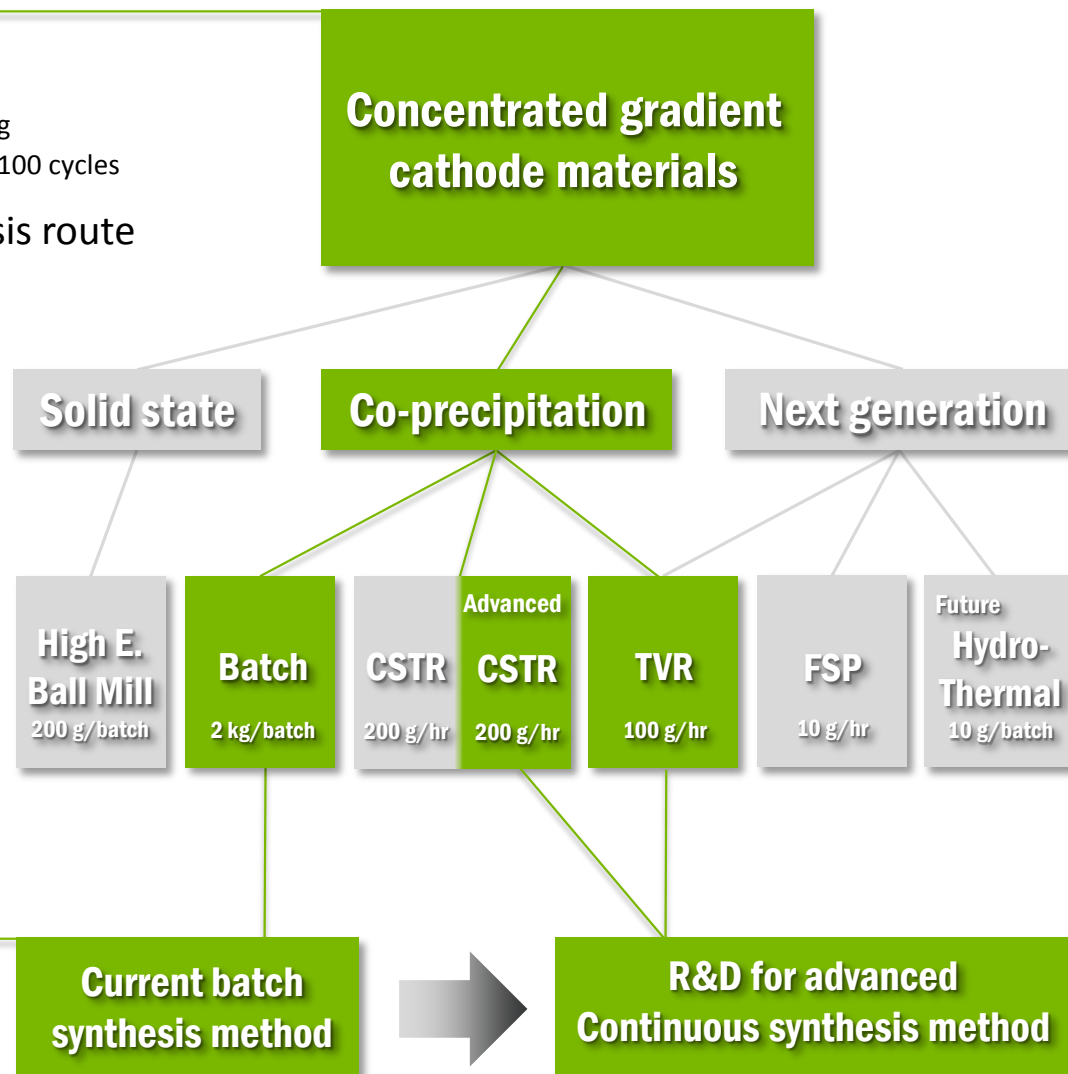
FY16	Material	Characterize preliminary 622 Gradient material	Completed	Q4
	Material	Deliver preliminary 622 Gradient material to BNL	Completed	
		– X-ray absorption spectroscopy		
FY17		– Thermal stability studies	Completed	Q1
	Material	Design high-capacity concentrated gradient materials	Completed	
		– 811 Gradient material ($\text{LiNi}_{0.90}\text{Mn}_{0.05}\text{Co}_{0.05}\text{O}_2$ core + $\text{LiNi}_{0.33}\text{Mn}_{0.33}\text{Co}_{0.33}\text{O}_2$ surface)		
		– 8 μm particle structure design (Core-Gradient & Core-Shell)		
	Process	Set up 20L batch reactor system for material synthesis	Completed	
	Material	Synthesize preliminary 811 Core-Gradient material	Completed	Q2
	Material	Synthesize preliminary 811 Core-Shell material	Completed	
	Material	Characterize two prepared gradient materials	Completed	
		– Cross-sectional mapping (SEM with EDS)		
		– Electrochemical test and comparison		
		– Thermal stability studies	Planned	Q3
	Material	Kg production of 811 Core-Shell and Core-Gradient materials	Planned	
	Material	Design other types of concentrated gradient materials such as FCG* and TSFCG**	Planned	
Process	Investigate an advanced continuous synthesis process using 10L TVR***	Planned		

* Full Concentration-Gradient ** Two Slope Full Concentration-Gradient *** Taylor Vortex Reactor

Approach - Strategy

Material Synthesis and Evaluation

- ❑ Define target active material
 - Concentrated gradient cathode materials targeting > 220 mAh/g with > 95% capacity retention after 100 cycles
- ❑ Select synthesis process and synthesis route
 - Current: Batch process with hydroxide route
 - Future: Advanced continuous TVR process
- ❑ Produce preliminary materials
 - Gram scale preliminary synthesis of various types of concentrated gradient materials
 - Performance check for Go/No-Go decision
- ❑ Investigate particle structure and composition combination
- ❑ Production and distribution
 - 1 ~ 10 kilogram scale production
 - Characterization & pouch cell evaluation
 - Material support to assist other DOE programs






Process R&D for Scale-up

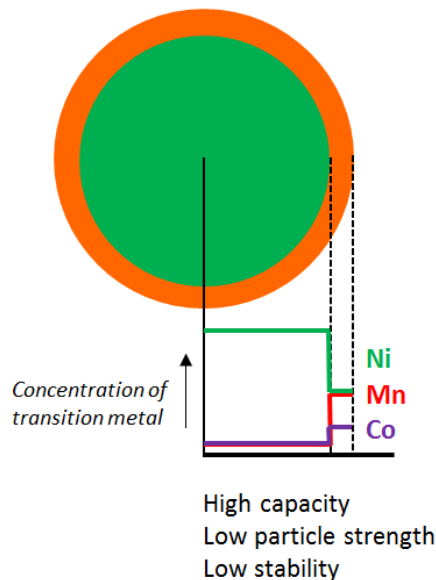
- ❑ Current 40L batch process
- ❑ Future continuous process

Gradient Material Strategy

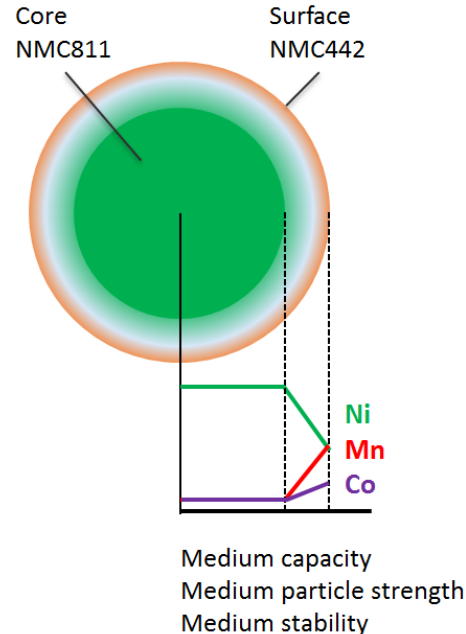
- Gradient material will have the best of Core and Surface compositions

	Core composition	- Ni-rich material : high capacity, low stability
	Gradient layer	- Gradient layer : prevents the crack and segregation between Core and Shell
	Surface composition	- Mn-rich material : low capacity, high stability

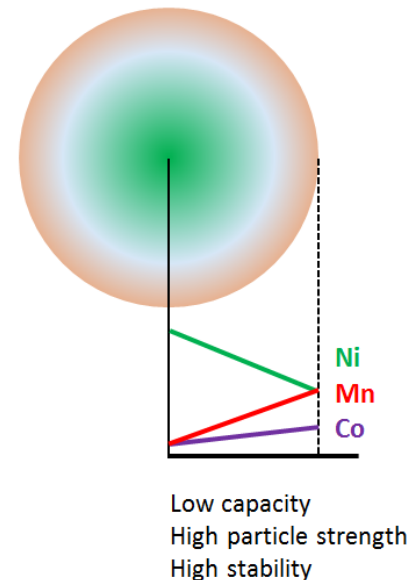
[GEN 1] Core-Shell



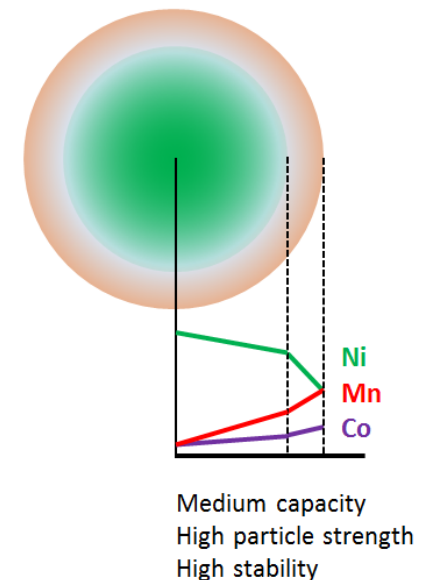
[GEN 2] Core-Gradient (FY16)



[GEN 3] Full Concentration-Gradient



[GEN 4] Two Slope Full Concentration-Gradient

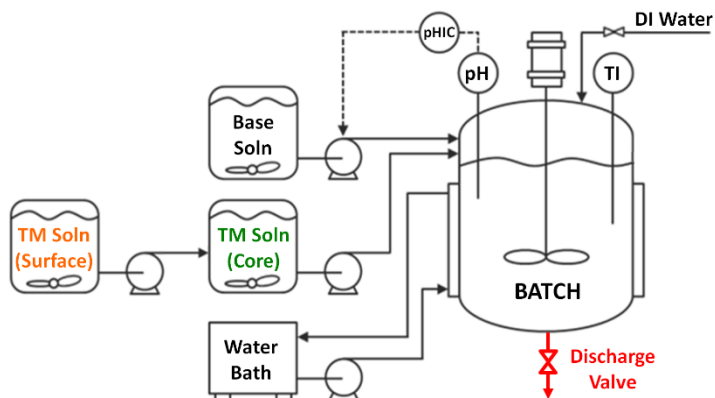


Research target

- 1 Higher capacity - increase Ni portion
- 2 Higher particle strength – particle structure optimization
- 3 Higher stability – composition optimization

Gradient Material Synthesis Process

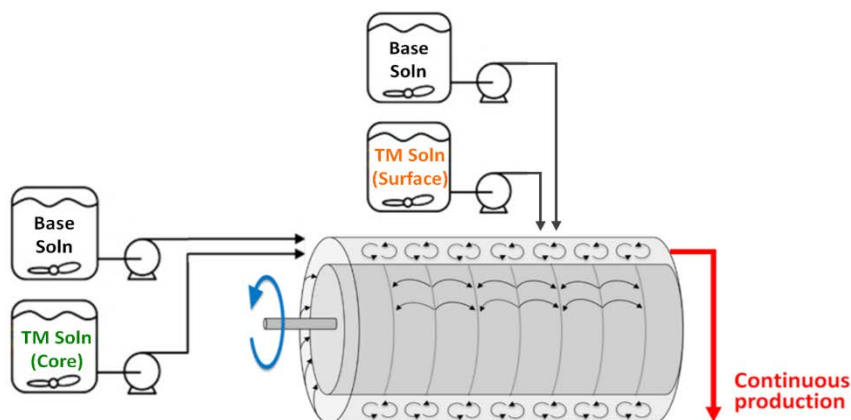
- Batch reactor system was used for gradient material synthesis *Current batch production*



- * First, Core TM solution feeding to batch reactor
Then, Surface TM solution feeding to Core TM solution tank

✓ Core TM solution changes to Surface TM solution gradually

- Continuous production of gradient material is being investigated using Taylor vortex reactor system *Process development for continuous production*



✓ Surface TM solution is injected to the middle point of TVR

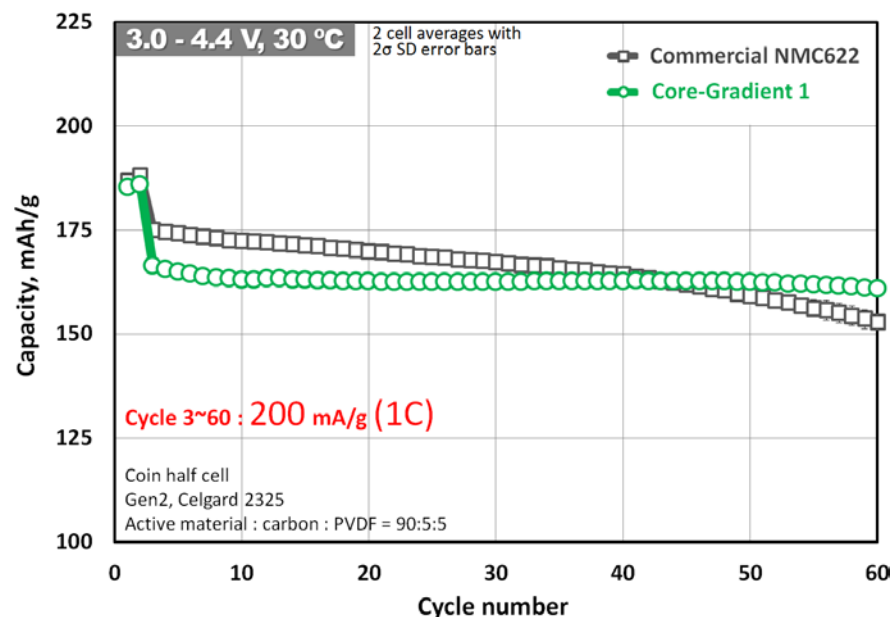
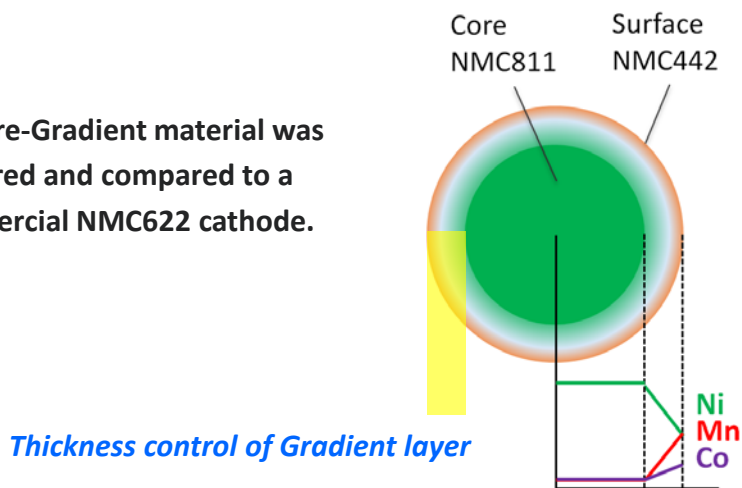


10 L Taylor Vortex Reactor in place

622 Gradient Material (FY16)

□ Synthesis of 622 Gradient materials

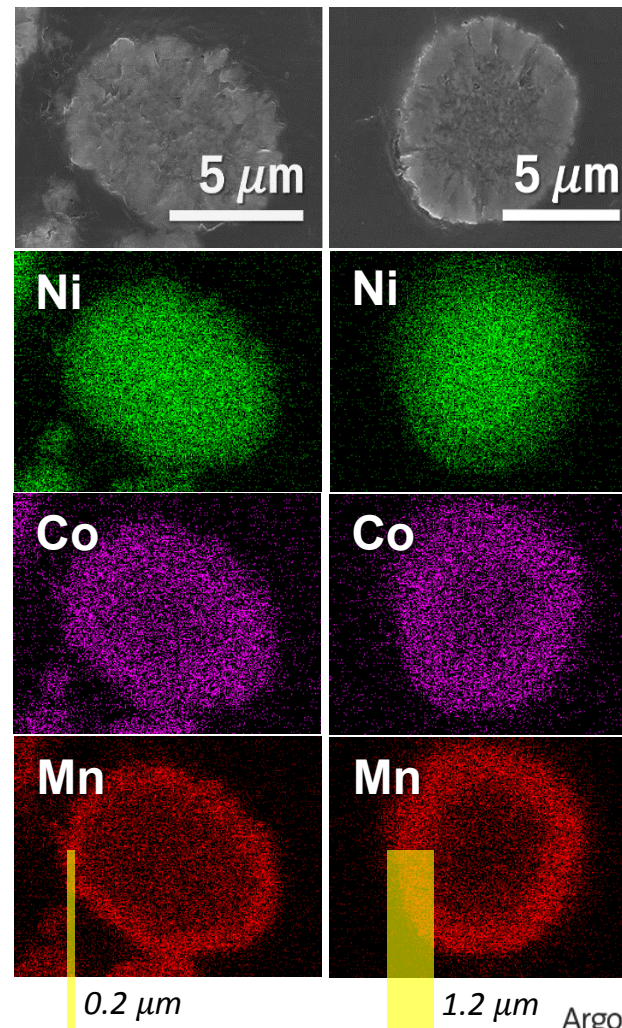
■ Core-Gradient material was prepared and compared to a commercial NMC622 cathode.



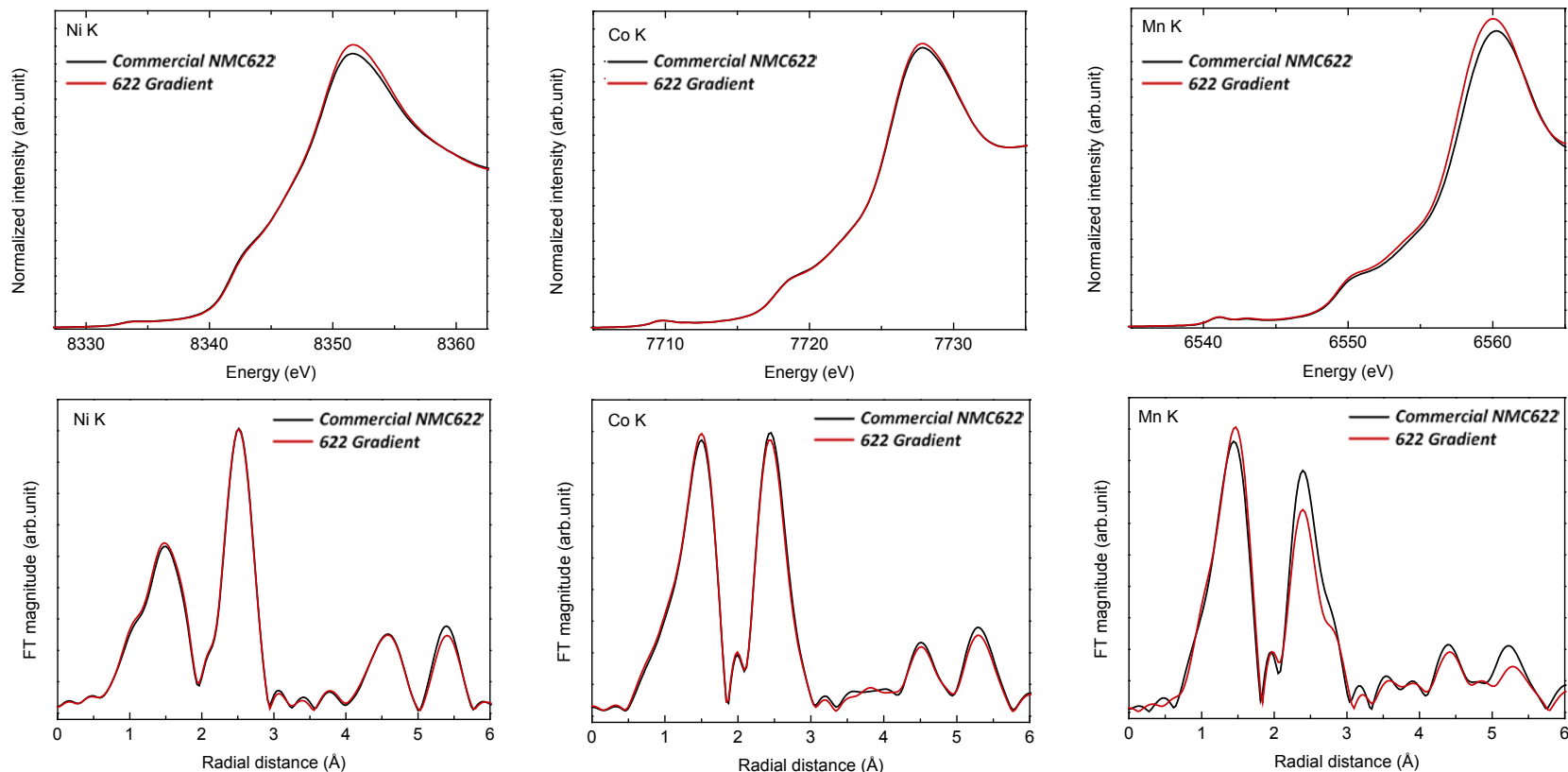
□ Elemental mappings

Core-Gradient 1
(thin layer)

Core-Gradient 2
(thick layer)



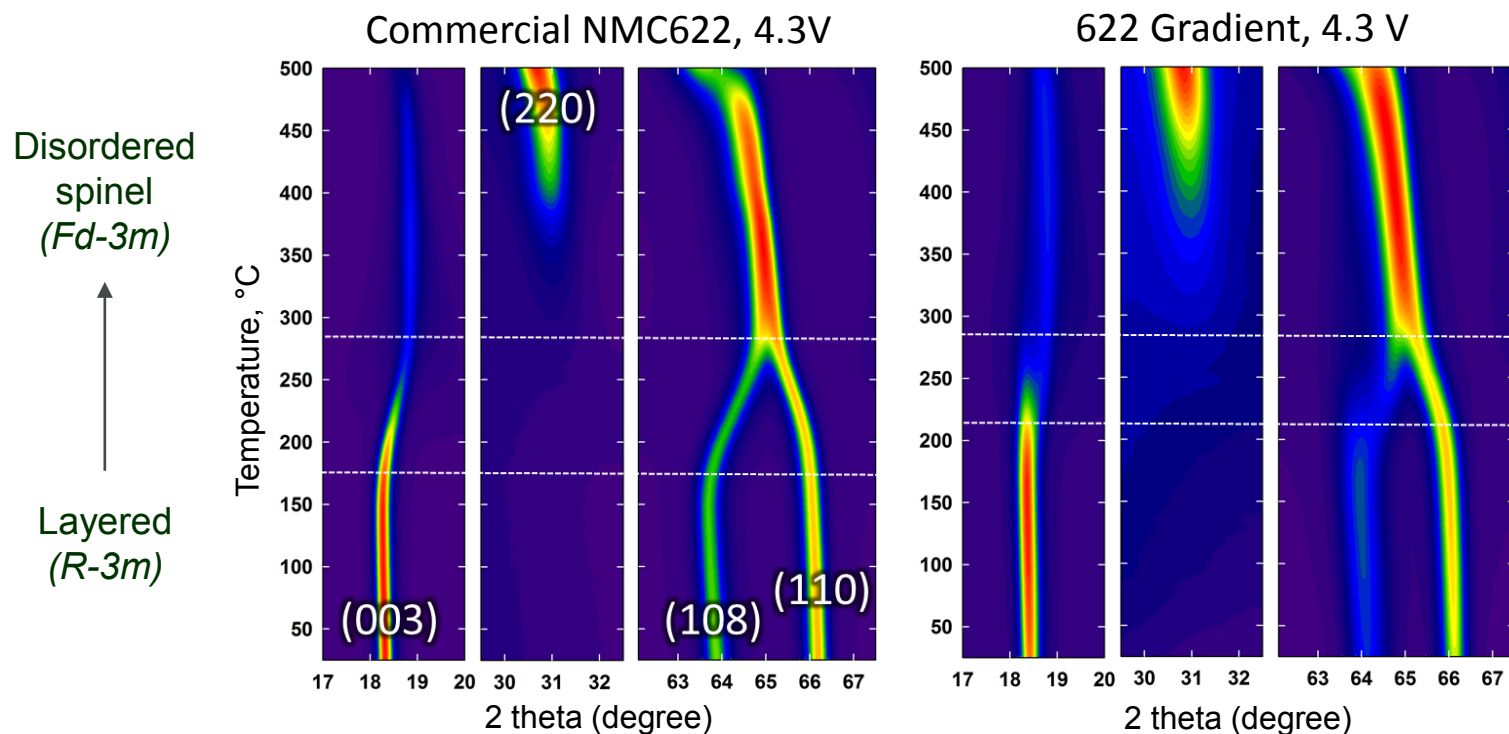
- X-ray absorption spectroscopy on commercial NMC622 and 622 Gradient materials
(BNL, EES, Dr. Xiao-Qing Yang's group)



- ✓ The Ni, Co and Mn K-edge XANES data are almost identical, which indicates that the average oxidation states of Ni, Co and Mn in commercial NMC622 and 622 Gradient cathodes are same.
- ✓ The local structure of Ni and Co are same. But Mn-M (M=Ni, Co, Mn) correlation at ~2.5 Å shows a difference. The local structural analysis near surface by using soft X-ray and TEM will be conducted.

622 Gradient Material Characterization

- Thermal stability studies on the charged 622 Gradient using time resolved XRD
(BNL, EES, Dr. Xiao-Qing Yang's group)



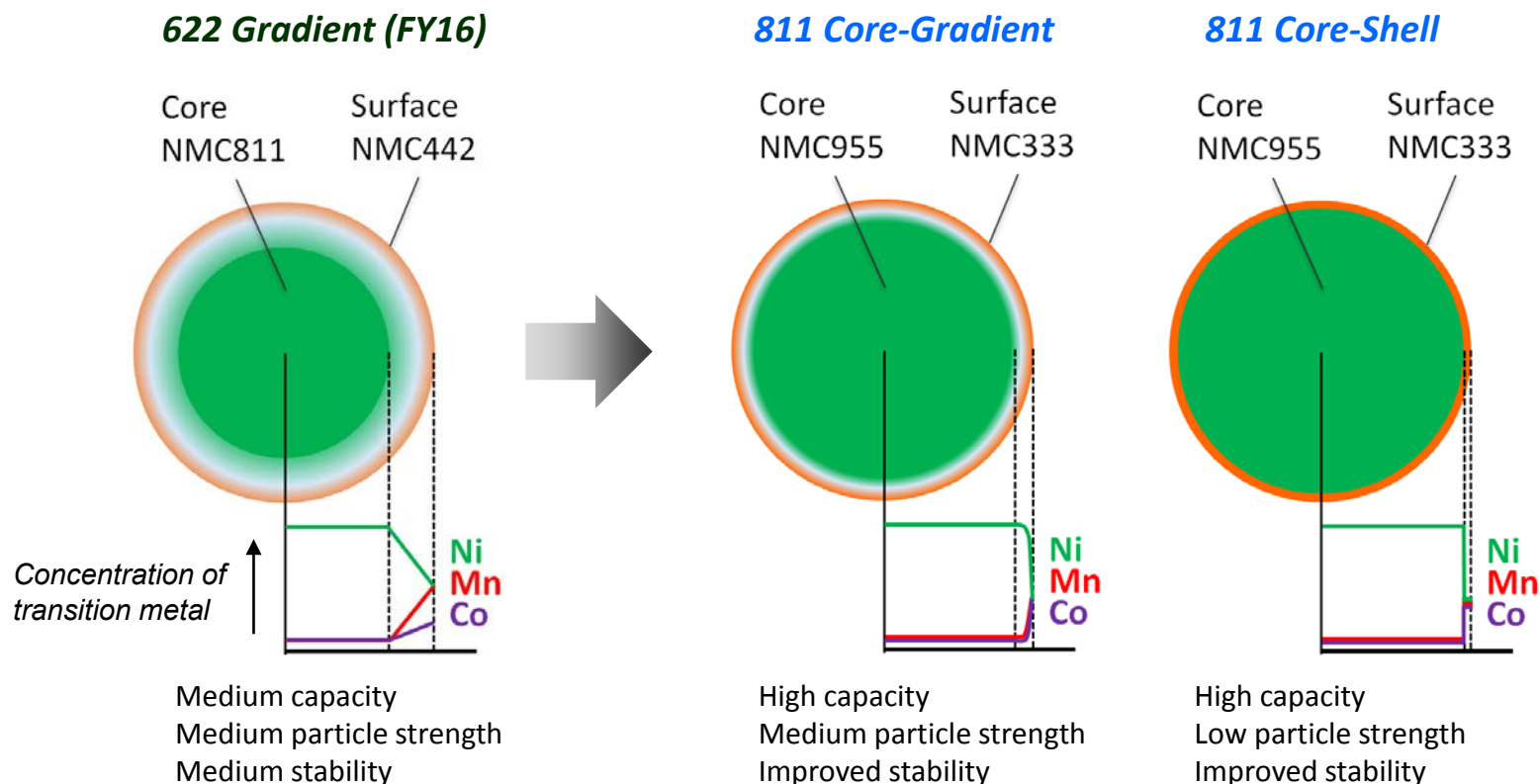
- ✓ The first phase transition from layered to spinel started at 175°C for commercial NMC622 while 622 Gradient shows the phase transition from at about 200°C.
- ✓ Peak broadening at charged state for 622 Gradient indicates the structural inhomogeneity characteristic.
- ✓ The gas analyze will be conducted to investigate the difference in (003) peak shift behavior.

811 Gradient Material Synthesis

❑ To achieve higher-capacity gradient material with better cycle life

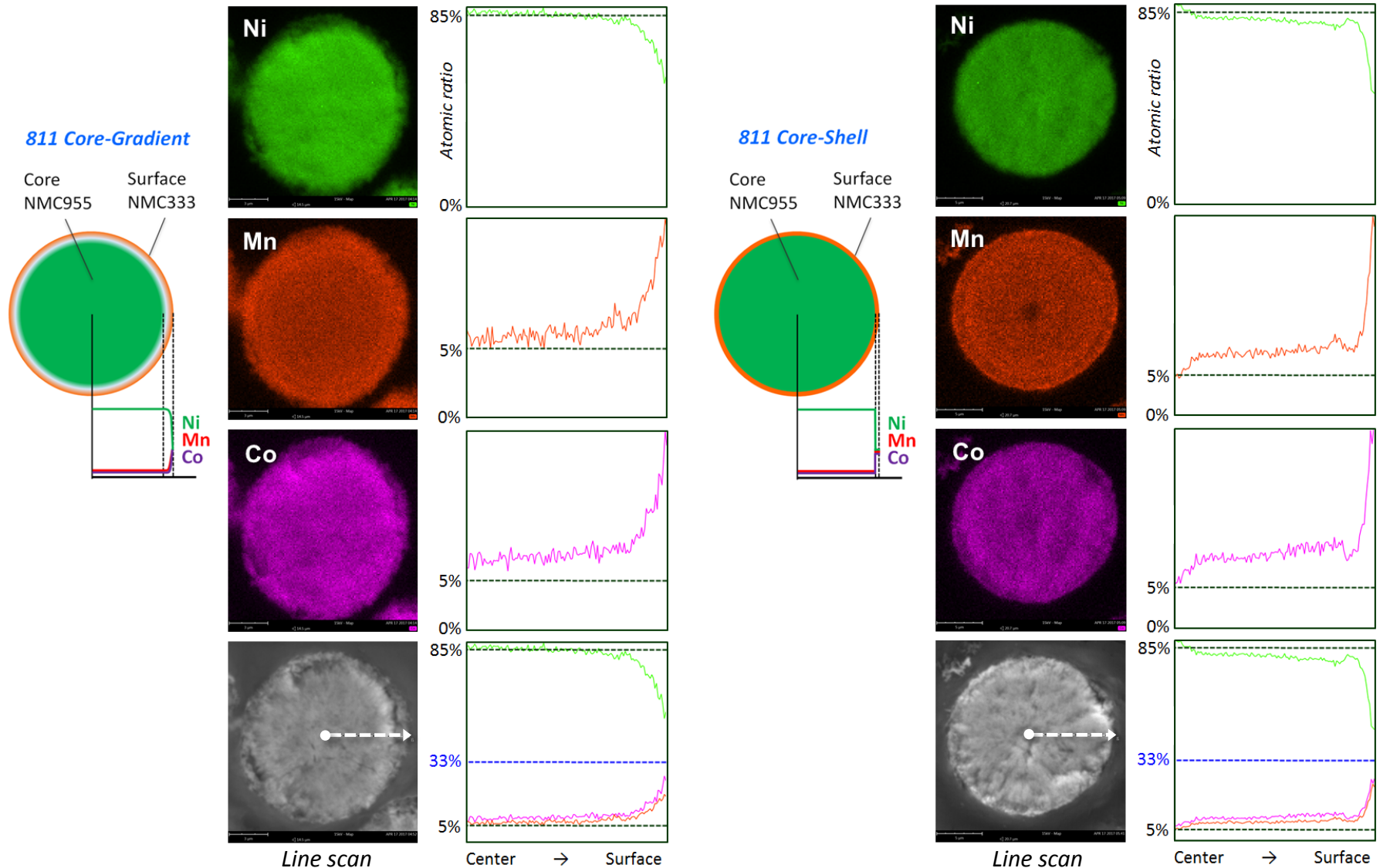
- 1 Core composition – $\text{LiNi}_{0.90}\text{Mn}_{0.05}\text{Co}_{0.05}\text{O}_2$ for higher capacity
- 2 Surface composition – $\text{LiNi}_{0.33}\text{Mn}_{0.33}\text{Co}_{0.33}\text{O}_2$ for better stability
- 3 Overall composition – $\text{LiNi}_{0.80}\text{Mn}_{0.10}\text{Co}_{0.10}\text{O}_2$

❑ Particle structure design for 811 Gradient materials



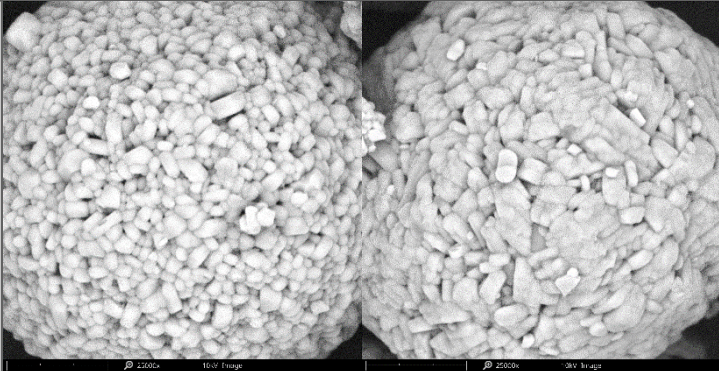
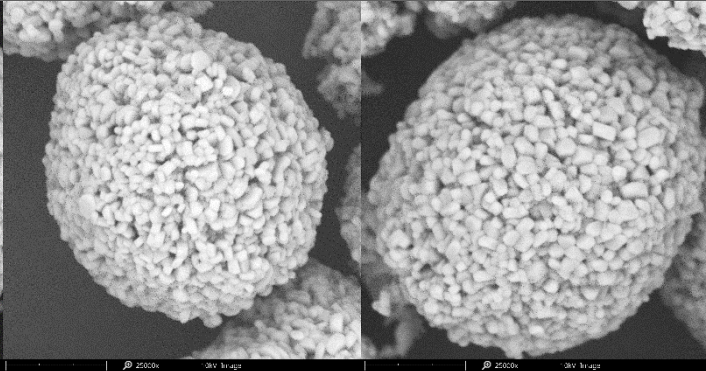
811 Gradient Material Synthesis

□ SEM with EDS on synthesized 811 Gradient materials



Commercial NMC and 811 Gradient Materials

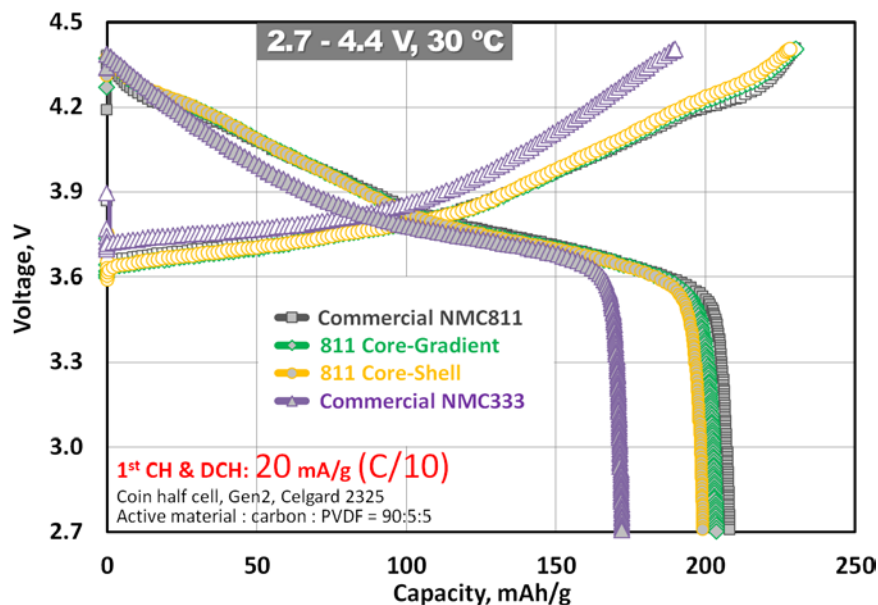
□ Comparison of 811 Gradient material prepared using batch reactor

Material	NMC 811	NMC 333	811 Core-Gradient	811 Core-Shell
Scale / status	Commercial product		MERF pre-pilot preliminary product	
SEM				
Composition	NMC 811	NMC 333	~ NMC 811	~ NMC 811
ICP-MS analysis	$\text{Li}_{1.04}\text{Ni}_{0.80}\text{Mn}_{0.10}\text{Co}_{0.10}\text{O}_y$	$\text{Li}_{1.07}\text{Ni}_{0.34}\text{Mn}_{0.33}\text{Co}_{0.33}\text{O}_y$	$\text{Li}_{1.00}\text{Ni}_{0.76}\text{Mn}_{0.12}\text{Co}_{0.12}\text{O}_y$	$\text{Li}_{1.00}\text{Ni}_{0.76}\text{Mn}_{0.12}\text{Co}_{0.12}\text{O}_y$
Particle size D_{50} [μm]	13.7	11.7	8.0	8.0
BET [m^2/g]	0.37	0.34	0.55	0.57
* FCE [%]	90.0	90.5	88.1	86.7
* Avg. working voltage	3.86	3.88	3.87	3.87
* Discharge capacity @ 5 cycle [mAh/g]	210.0	171.6	209.1	205.4

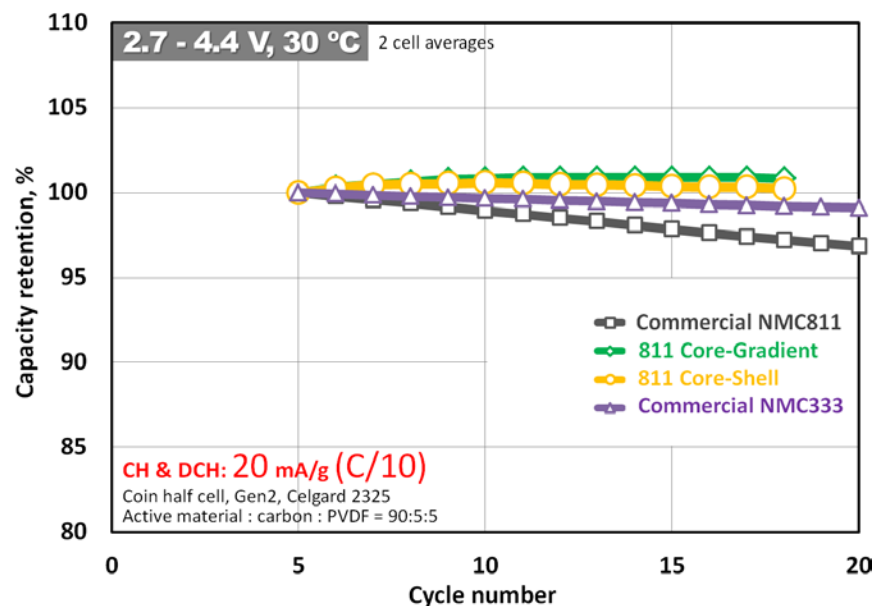
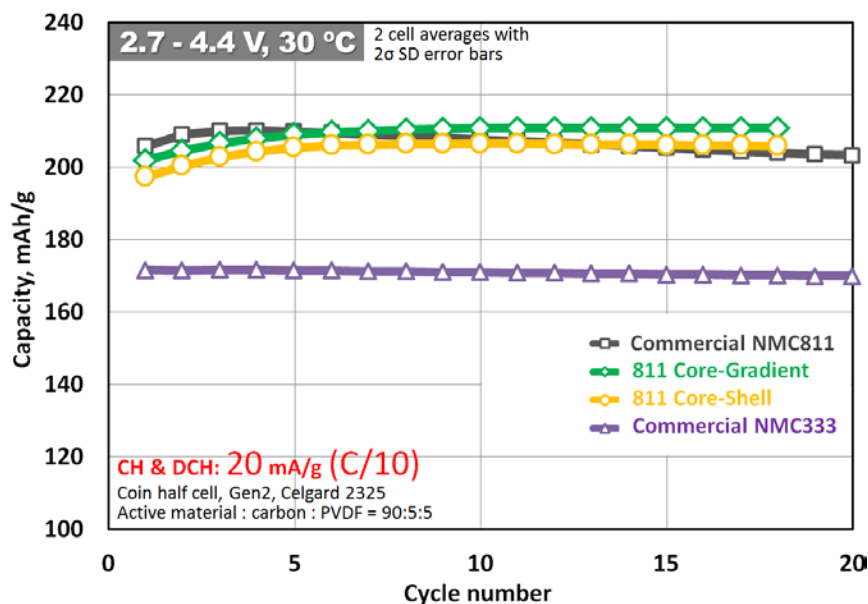
* At C/10, 2.7 – 4.4 V and 30°C

✓ Gradient materials show reasonable physical electrochemical properties compared to a commercial NMC811.

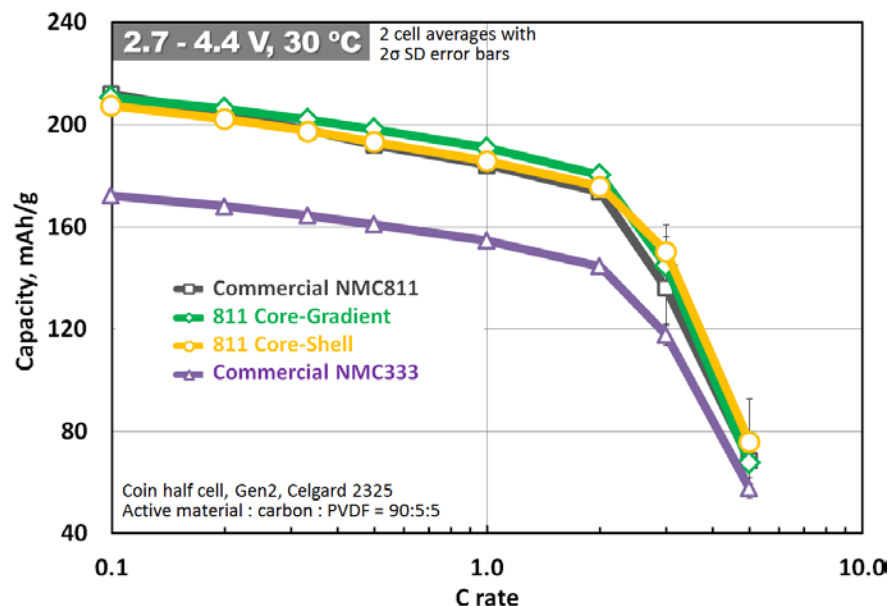
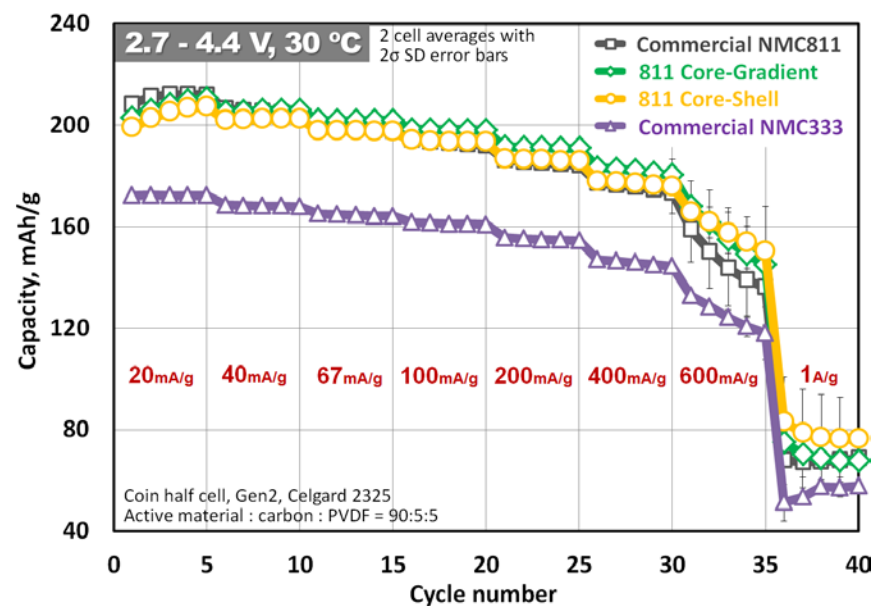
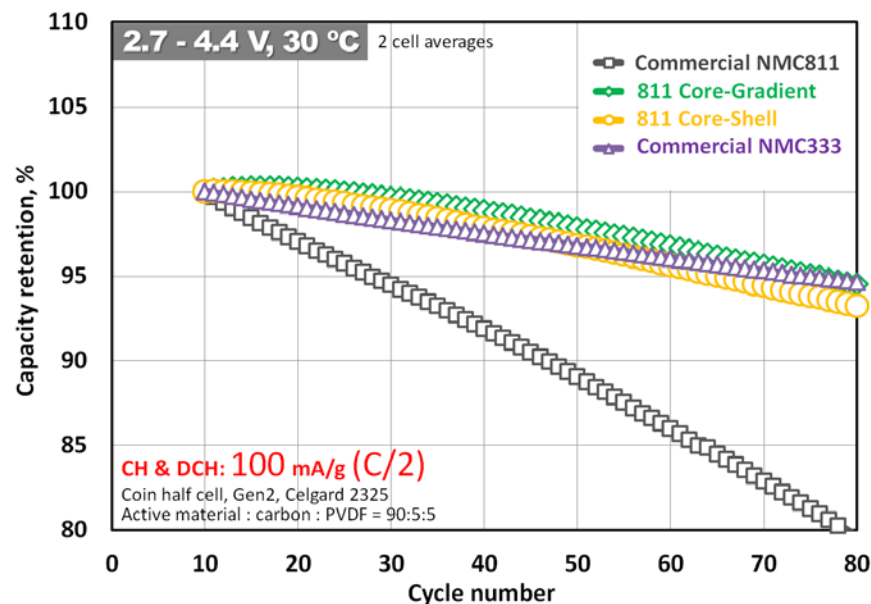
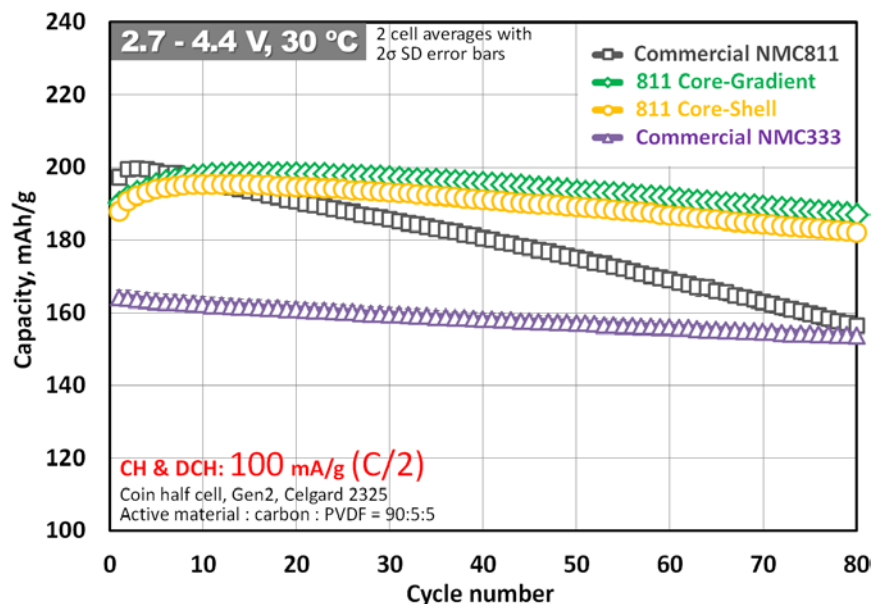
30°C Voltage Profile and C/10 Cycling



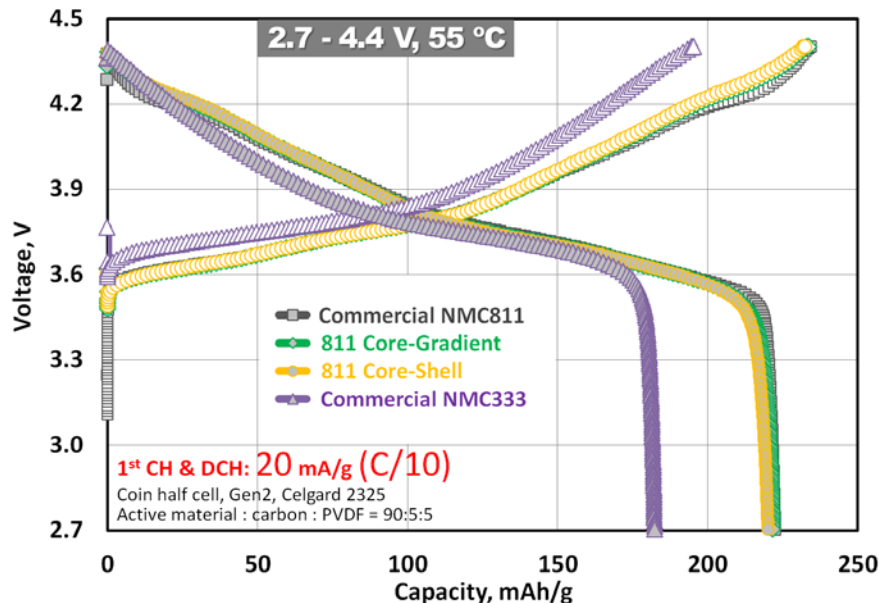
- ✓ Both Core-Gradient and Core-Shell show high initial discharge capacity of 210 mAh/g at C/10 and 30°C.
- ✓ Gradient materials show improved capacity retention compared to commercial NMC811 and NMC333 at C/10 and 30°C.



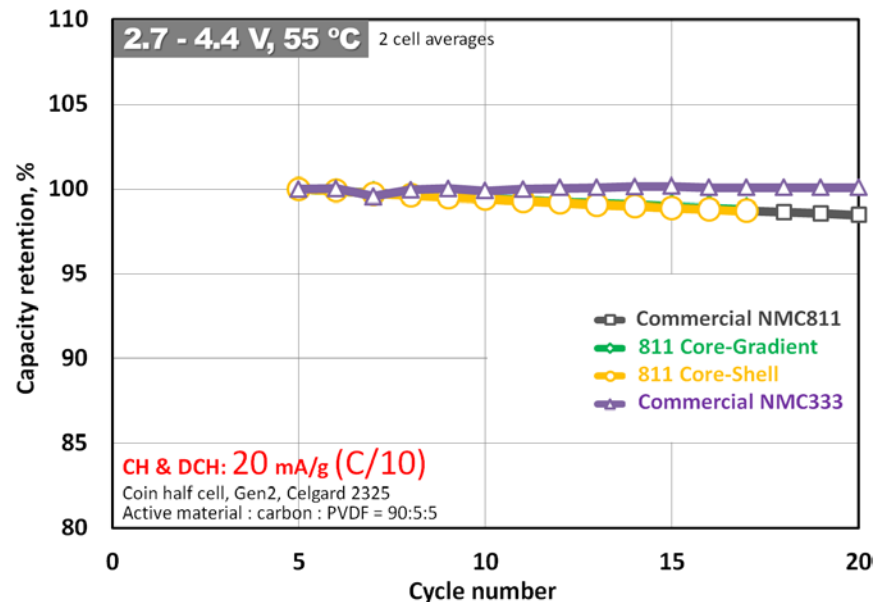
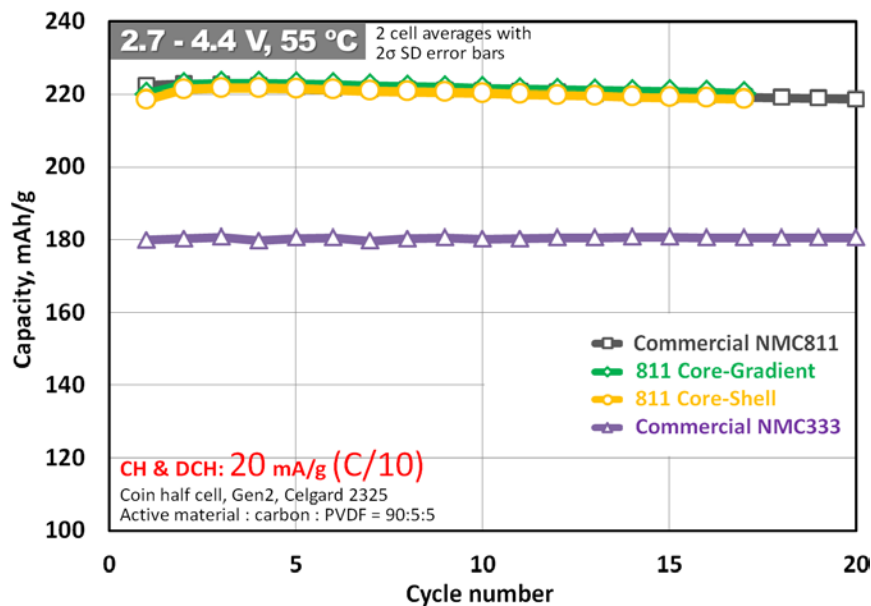
30°C C/2 Cycling and Rate Performance



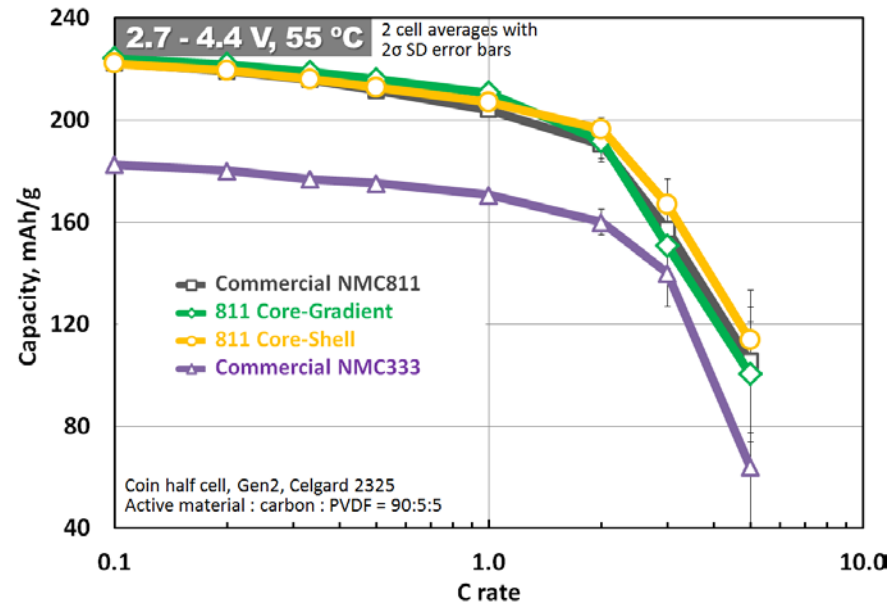
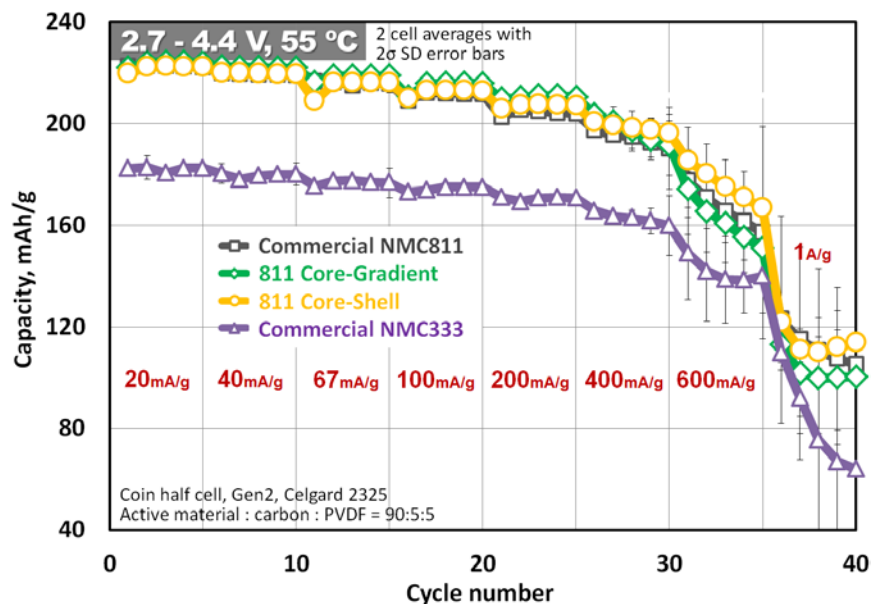
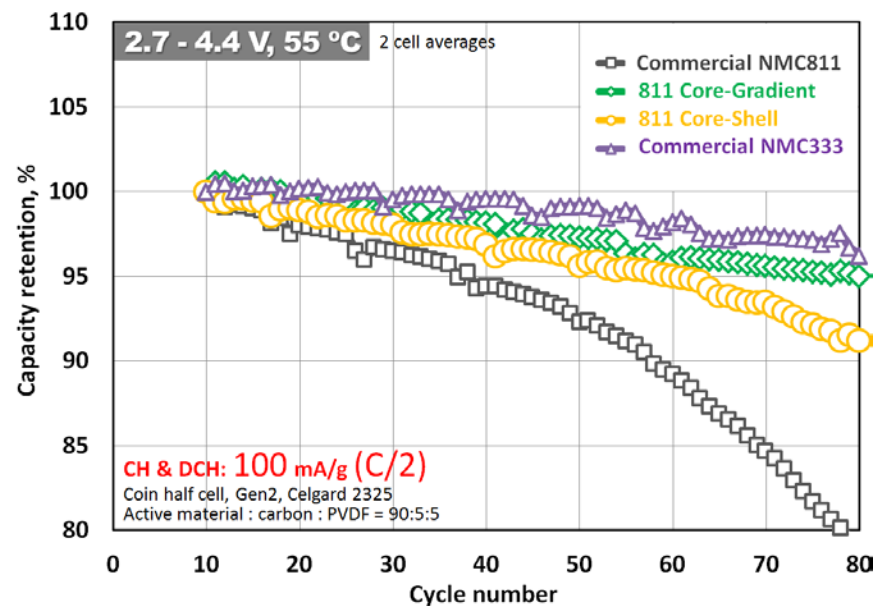
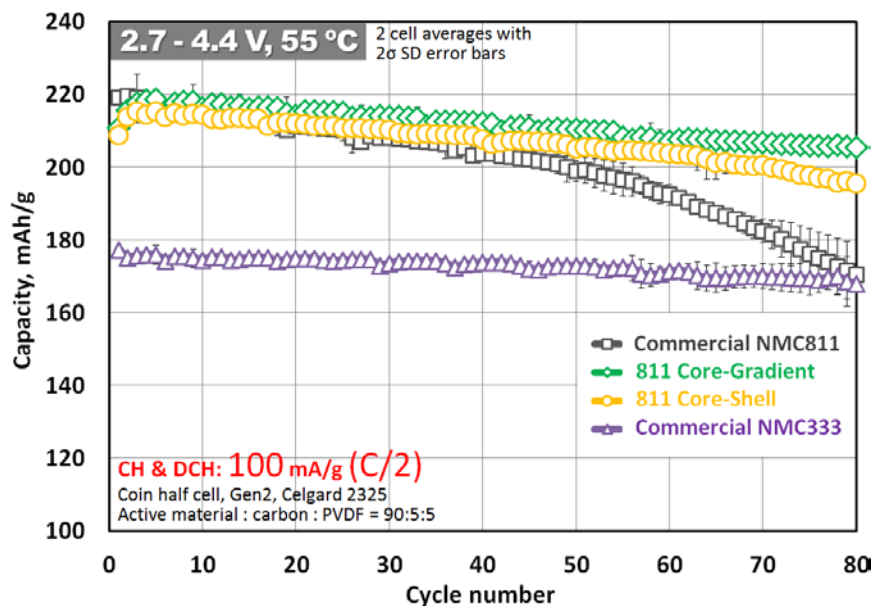
55°C Voltage Profile and C/10 Cycling



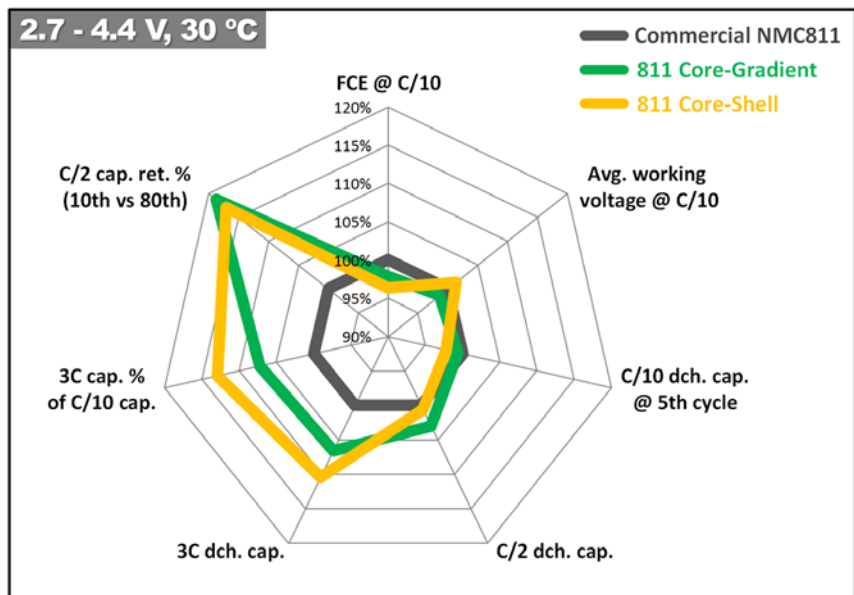
- ✓ Both Core-Gradient and Core-Shell show high initial discharge capacity of 220 mAh/g at C/10 and 55°C.
- ✓ Gradient materials show similar capacity retention compared to commercial NMC811 at C/10 and 55°C.



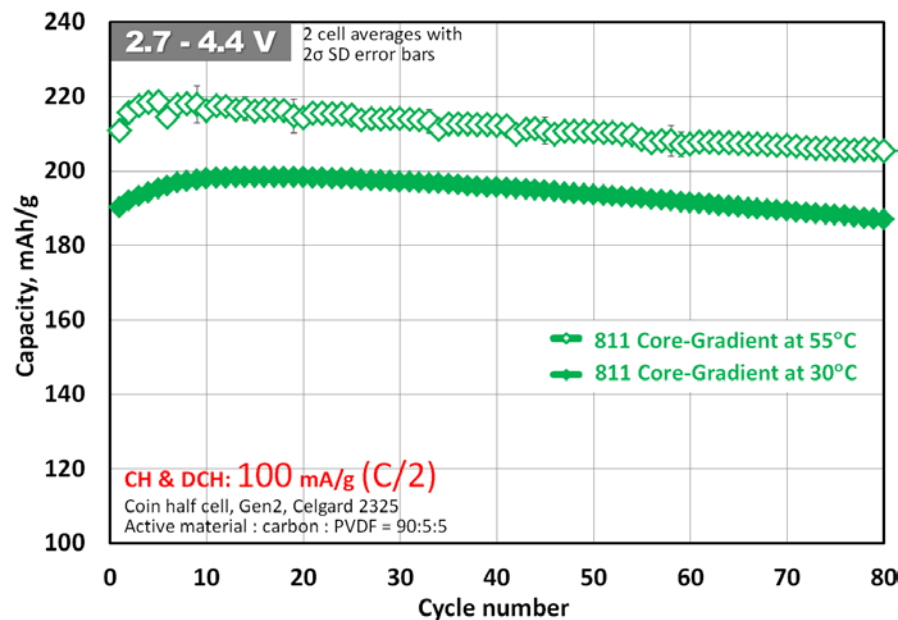
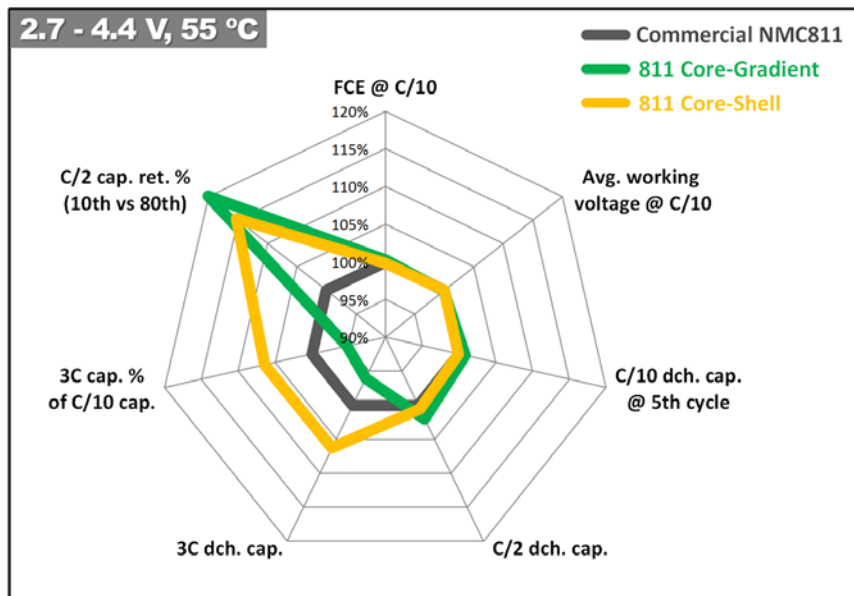
55°C C/2 Cycling and Rate Performance



Radar Map Comparison and Conclusion



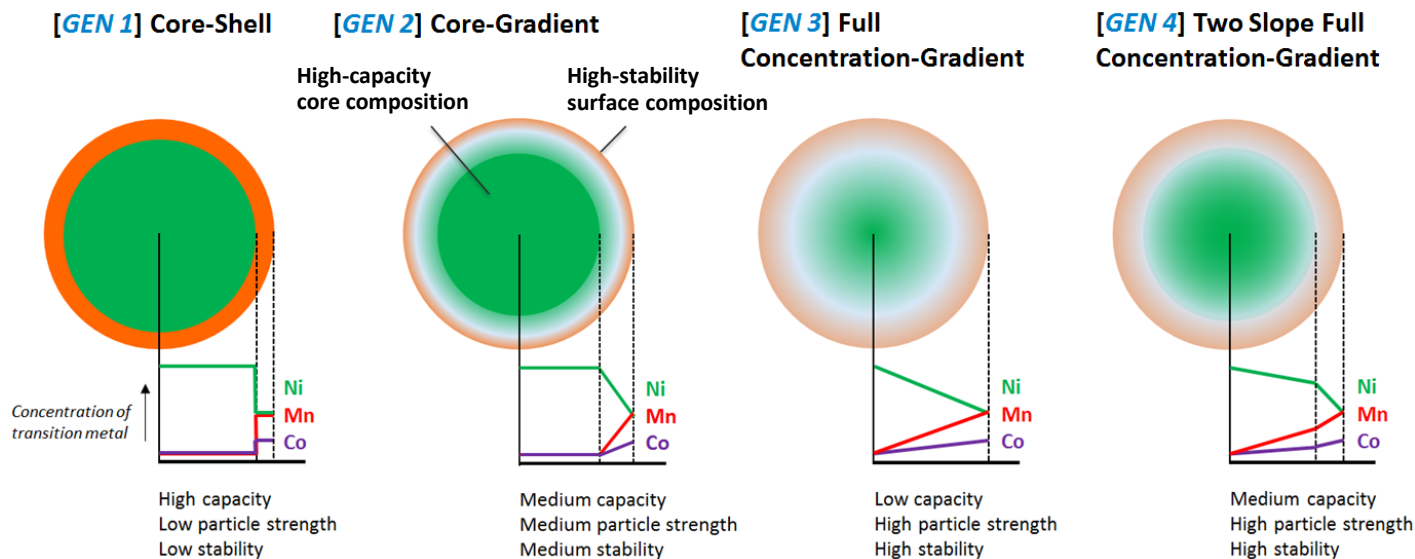
- ✓ 811 Core-gradient material shows 20% improved capacity retention compared to a commercial NMC811.
- ✓ 811 Core-Shell material shows better rate capability.
- ✓ To further improve cycle life of NMC gradient materials, other additional approaches are necessary such as surface coating.



Long-term Research Plan for Gradient Materials

Investigate various particle structures and composition combinations

- Particle strength analysis
- 3D elemental mapping
- Thermal stability



Investigate an advanced continuous synthesis process

- Advanced CSTR (Continuous Stirred Tank Reactor) system
- TVR (Taylor Vortex Reactor) system

Surface coating research on 811 Core-Gradient material to achieve improved cycle life

Responses to Previous Year Reviewers' Comments

- “The reviewer remarked outstanding performance meeting the needs of the industry and research community while adding value to ANL’s thorough licensing strategies.”
 - Response: *We will try to invent new types of concentrated gradient cathode material and develop its customized synthesis process which will generate several material and/or process patents to strengthen ANL’s thorough licensing strategies.*

- “The reviewer said that the list of the collaborators demonstrated the trust this group has earned, well done. It will be interesting to see the performance of the commercialized products.”
 - Response: *The characterization of the synthesized materials will be carried out in collaboration with several basic research groups for fundamental understanding. The physical and electrochemical properties of the synthesized 811 Core-Shell and Core-Gradient materials were compared to commercial NMC811 and NMC333. As such, newly prepared concentrated gradient cathodes will constantly be compared to an available commercial products with statistical evaluation.*

- “The reviewer commented that proposed future research is well-balanced to meet the program objectives and said highly qualified, hard-working team.”
 - Response: *We will develop both new types of concentrated gradient cathode material and their new synthesis process which has economical feasibility with expertise and effort.*

Collaborations

- Material Characterization:
 - Brookhaven National Lab (Seongmin Bak)
 - Thermal stability
 - University of Illinois at Chicago (Jordi Cabana)
 - 3D elemental mapping
 - Technische Universität Braunschweig (Wolfgang Haselrieder)
 - Particle stress evaluation
- Synthesis process R&D:
 - Laminar Co., Ltd – CRADA
 - Taylor Vortex Reactor process scale-up
- Electrochemical evaluation of scaled materials:
 - Argonne's CAMP facility (Andrew Jansen)
- Thermal stability evaluation:
 - 622 Core-shell cathode to Brookhaven National Lab
 - 622 Core-gradient cathode to Brookhaven National Lab



Open to working with any group developing advanced active materials that will be beneficial for the ABR program.

Remaining Challenges and Barriers

- Development and scale-up of concentrated gradient cathode material is challenging but has great promise to improve the performance of battery materials.
- Continuous synthesis process for concentrated gradient cathode material need to be developed to lower manufacturing cost.
- Further improvement of concentrated gradient cathode material by surface coating is necessary to achieve longer cycle life.

Proposed Future Research

- **811 Core-Shell and Core-Gradient materials**
 - Kilogram scale-up of 811 Core-Shell and Core-Gradient materials (FY17)
 - Large format cell evaluation (FY17)
 - Thermal stability studies (FY17)
 - Surface coating studies to improve cycle life (FY17)
 - Particle strength analysis
 - 3D elemental mapping
- **Preliminary synthesis of other types of 811 concentrated gradient materials**
 - Investigate various particle structures such as FCG and TSFCG
 - Material characterization and comparison
- **Investigate an advanced continuous synthesis process**
 - Develop a customized continuous process for concentrated gradient materials

Any proposed future work is subject to change based on funding levels.

Summary

- **622 Gradient material characterization**
 - X-ray absorption spectroscopy
 - Thermal stability studies
- **811 Core-Shell and Core-Gradient materials**
 - Core ($\text{LiNi}_{0.90}\text{Mn}_{0.05}\text{Co}_{0.05}\text{O}_2$) and surface ($\text{LiNi}_{0.33}\text{Mn}_{0.33}\text{Co}_{0.33}\text{O}_2$) composition were determined
 - Particle structure (Core-Shell and Core-Gradient) was selected
 - 20L batch synthesis system was set up
 - Preliminary 811 Core-Shell material was synthesized
 - Preliminary 811 Core-Gradient material was synthesized
 - Cross-sectional mapping (SEM with EDS)
 - Electrochemical test and comparison to commercial products
- **Develop a customized continuous process**
 - Taylor Vortex Reactor system is being investigated

Acknowledgements and Contributors

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- **Technische Universität Braunschweig**
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